



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Table of Contents Crossword 11 18 14 Hss Physical Science

1. Understanding the eBook Crossword 11 18 14 Hss Physical Science
 - The Rise of Digital Reading Crossword 11 18 14 Hss Physical Science
 - Advantages of eBooks Over Traditional Books
2. Identifying Crossword 11 18 14 Hss Physical Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Crossword 11 18 14 Hss Physical Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from Crossword 11 18 14 Hss Physical Science
 - Personalized Recommendations
 - Crossword 11 18 14 Hss Physical Science User Reviews and Ratings
 - Crossword 11 18 14 Hss Physical Science and Bestseller Lists
5. Accessing Crossword 11 18 14 Hss Physical Science Free and Paid eBooks

- Crossword 11 18 14 Hss Physical Science Public Domain eBooks
 - Crossword 11 18 14 Hss Physical Science eBook Subscription Services
 - Crossword 11 18 14 Hss Physical Science Budget-Friendly Options
6. Navigating Crossword 11 18 14 Hss Physical Science eBook Formats
- ePub, PDF, MOBI, and More
 - Crossword 11 18 14 Hss Physical Science Compatibility with Devices
 - Crossword 11 18 14 Hss Physical Science Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Crossword 11 18 14 Hss Physical Science
 - Highlighting and Note-Taking Crossword 11 18 14 Hss Physical Science
 - Interactive Elements Crossword 11 18 14 Hss Physical Science
8. Staying Engaged with Crossword 11 18 14 Hss Physical Science
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Crossword 11 18 14 Hss Physical Science
9. Balancing eBooks and Physical Books Crossword 11 18 14 Hss Physical Science
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Crossword 11 18 14 Hss Physical Science
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Crossword 11 18 14 Hss Physical Science
- Setting Reading Goals Crossword 11 18 14 Hss Physical Science
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Crossword 11 18 14 Hss Physical Science
- Fact-Checking eBook Content of Crossword 11 18 14 Hss Physical Science
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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