

POGIL: Covalent Bonding

So far you have learned that atoms can bond via ionic bonding. However, they can also bond with covalent and metallic bonding. Today we will be taking a look at covalent bonding. Have out a periodic table, and let's begin.

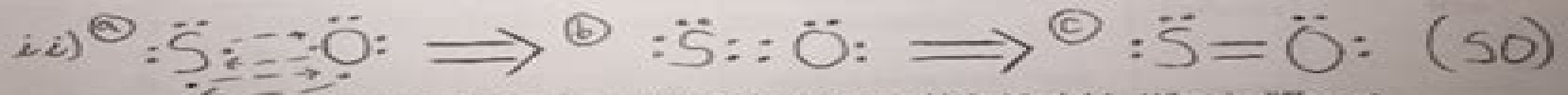
Model 1: Ionic Bonding with Lewis Dot Structures



Refer to Model 1 to answer questions 1-6:

1. What do the dots in Lewis dot structures represent (be specific)?
2. In step a, what is the dotted arrow indicating?
3. Do the ions in step b fulfill the octet rule?
4. What charge does the sodium ion result in getting? Write it on the dotted line in step b.
5. What charge does the chloride ion result in getting? Write it on the dotted line in step b.
6. Considering their charges, why do the sodium and chloride ions "desire" to bond, resulting in the ionic compound found in step c?

Model 2: Covalent Bonding with Lewis Dot Structures



7. Look at steps a and b in Model 2 i, and compare it to steps a and b in Model 1. What is different about the movement of electrons in Model 2 i?
8. Is your answer to #7 also true in regards to Model 2 ii?
9. Thinking about your answers to #'s 7 and 8, why don't sulfur and oxygen get charges?

Covalent Bonding Pogil

R Sanderson



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POGIL Shawn R. Simonson, 2023-07-03 Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project *Organic Chemistry* Suzanne M. Ruder, The POGIL Project, 2015-12-29 ORGANIC CHEMISTRY

Russian Journal of Coordination Chemistry , 1998 *The Nature of the Chemical Bond and the Structure of Molecules and Crystals* Linus Pauling, 1960 Thorough discussion of the various types of bonds their relative natures and the structure of molecules and crystals *The Covalent Bond* Henry Sinclair Pickering, 1977 *Chemical Bonding and the Geometry of Molecules* George E. Ryschkewitsch, 1963 **Chemistry of the Covalent Bond** Leallyn Burr Clapp, 1957 **Structure and Bonding** Jack Barrett, 2001 This book explains in non mathematical terms where possible the factors that govern

covalent bond formation the lengths and strengths of bonds and molecular shapes **The Chemical Bond** Gernot Frenking, Sason Shaik, 2014-06-13 This is the perfect complement to *Chemical Bonding Across the Periodic Table* by the same editors who are two of the top scientists working on this topic each with extensive experience and important connections within the community The resulting book is a unique overview of the different approaches used for describing a chemical bond including molecular orbital based valence bond based ELF AIM and density functional based methods It takes into account the many developments that have taken place in the field over the past few decades due to the rapid advances in quantum chemical models and faster computers Chemical Bonds and Bonds Energy R Sanderson, 2012-12-02 *Chemical Bonds and Bonds Energy* Second Edition provides information pertinent to the fundamental aspects of contributing bond energy and bond dissociation energy This book explores the values that are useful in the interpretation of significant phenomena such as product distribution and reaction mechanisms Organized into 12 chapters this edition begins with an overview of the quantitative relationship among three basic properties of an atom namely nonpolar covalent radius electronegativity and homonuclear single covalent bond energy This text then examines the quantitative means of evaluating the partial atomic charges that result from initial differences in the electronegativity of atoms that form a compound Other chapters consider the recognition of the reduction of bond weakening not by multiplicity and in certain types of single covalent bonds The final chapter deals with the application of the principal ideas and techniques to the oxidation of ethane This book is a valuable resource for organic and inorganic chemists Chemical Bonding M.S. Sethi & M. Satake, 2010 Contents *Chemical Bonding I Basic Concepts* *Chemical Bonding II Additional Aspects* Intermolecular Force and Crystal Structures Covalent Bonding in Crystals, Molecules, and Polymers James C. Phillips, 1969 **Chemical Bonding** Kok Koon Leong, 2021-05-03 This book introduces the principles behind chemical bonding to teenagers between the ages of fifteen to seventeen Topics covered include ionic bonding covalent bonding and metallic bonding **Polar Covalence** R Sanderson, 2012-12-02 *Polar Covalence* provides a detailed account of a successful approach to understanding chemistry from knowledge of atomic structure and the properties that result from this structure This book discusses the nature of multiple bonds Organized into 16 chapters this book begins with an overview of the interrelationships of various basic atomic properties This text then describes chemical bonding which can only occur when the nuclei of both atoms can attract the same electrons Other chapters consider the bond energy of multiple bonds which can be determined by calculating the energy in the usual way as though the bonds were single but of the experimental length This book discusses as well the reduction of the lone pair bond weakening effect through the formation of multiple bonds The final chapter deals with the relative roles of principles and practice in the teaching of inorganic and general chemistry This book is a valuable resource for chemists and students **Chemical Bonds and Bond Energy** Robert Thomas Sanderson, 1971 *Teaching Chemical Bonding* Margaret Irene Lindsay, 1995 This document presents an instructional strategy for teaching chemical bonding using

parables and music Games student interactions and worksheets are included in the lesson plans Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding JRH

Chemical Binding and Structure J. E. Spice, 2016-01-22 Chemical Binding and Structure describes the chemical binding and structure in terms of current chemical theory This book is composed of 13 chapters and starts with a presentation of the principles of the old and modified quantum theory and its application The next chapters cover some basic topics related to chemical binding and structure including electrons the periodic table the electrovalent and covalent bonds and molecular geometry These topics are followed by discussions on the nature of the bond in transition metal complexes electronic and crystal structure crystallinity and other states of matter The concluding chapters are devoted to some analytical techniques for structure determination such as diffraction and spectroscopic methods This book is of value to high school and college chemistry teachers and students

Atoms & Chemical Bonding Science Learning Guide NewPath Learning, 2014-03-01 The Atoms Atomic Configuration Chemical Bonding Ionic Bonding Ionic Compounds Covalent Bonding Covalent Compounds Naming Compounds and Metallic Bonding Aligned to Next Generation Science Standards NGSS and other state standards

Chemistry of Chemical Bonding R. K. Sharma, 2007

The Concept of the Chemical Bond Zvonimir B. Maksic, 1990-06-13 The state of the art in contemporary theoretical chemistry is presented in this 4 volume set with numerous contributions from the most highly regarded experts in their field It provides a concise introduction and critical evaluation of theoretical approaches in relation to experimental evidence

Covalent Bonding Pogil Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Covalent Bonding Pogil**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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