

**NOTE:** *Diagrams are not drawn to scale. Show all work to receive credit.*

In problems 1 - 7 use the diagram below. Find each measure requested.

1. If  $JK = 13.7$ , find  $MN$
2. If  $MN = 22.8$ , find  $JK$
3. If  $JK = 2x + 4$  and  $MN = 36$ , find  $x$
4. If  $JK = [3(x + 2) + 6]$  and  $MN = 8(x + 1)$ , find  $JK$
5. If  $LJ = 3.7$ , find  $LM$
6. If  $LN = 15$ , find  $LK$
7. If  $LJ = 5x + 8$ ,  $JM = 9x$ , and  $KN = 7x + 6$ 
  - a. Find  $x$
  - b. Find  $LJ$
  - c. Find  $LN$



Use this figure for 1 - 7

In problems 8 - 9 use  $\triangle ABC$ , where  $D$ ,  $E$ , and  $F$  are the midpoints of the sides.

8. If  $DF = 2x + 10$  and  $AB = 5x + 8$ , find  $AE$
9. If  $AD = 3x$ ,  $DE = 2x + 1$ ,  $CB = 5x - 3$ , find  $AD$



Use this figure for 8 - 9

In problems 10 - 13 use the diagram below. Find each measure requested.

10. If  $DC = 18.4$ , find  $AD$ .
11. If  $AD = 5x$  and  $CD = 2x + 93$ , find  $CD$
12. If  $AD = 17$  and  $CD = 17$  and  $m\angle ABC = 46^\circ$ , find  $m\angle ABD$
13. If  $AD = 4$  and  $CD = 4$  and  $m\angle CBD = 28^\circ$ , find  $m\angle ABC$



Use this figure for 10 - 13

In problems 14 - 16 use the diagram shown to find each measurement requested.

14. Given that  $LT = 27.4$ ,  $UD = 23.8$  and  $DT = 27.4$ , find  $LD$ .
15. Given that  $m$  is the perpendicular bisector of  $\overline{LD}$ , and  $TD = 13.1$ , find  $LT$
16. Given that  $m$  is the perpendicular bisector of  $\overline{LD}$ , and  $TD = 3(x + 8)$  and  $LT = 5x$ , find  $LT$



Use this figure for 14 - 16

# Practice 8 1 Answer Key Geometry

**Taqdir Husain**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web jul 3 2019 a covalent bond in chemistry is a chemical link between two atoms or ions in which the electron pairs are shared between them a covalent bond may also be termed a molecular bond covalent bonds form between two nonmetal atoms with identical or relatively close electronegativity values this type of bond may also be found in other

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web important key terms to practice from chapter 6 terms in this set 20 covalent bond a bond formed when atoms share one or more pairs of electrons molecular orbital the region of high probability of finding an electron that is being shared between two nuclei in a covalent bond bond length

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web the carbon hydrogen chains in vulcanized rubber are held together by two sulfur atoms that form covalent bonds between the chains these covalent bonds are commonly called disulfide bridges explore other molecules that have such disulfide bridges present your findings to the class

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web because of the fact that polar covalent bonds have higher melting and boiling points why do dipole forces have an impact on the melting and boiling points of substances this is because the dipole forces are an added force of attraction that slows the molecules down making it take more energy to melt or boil them