

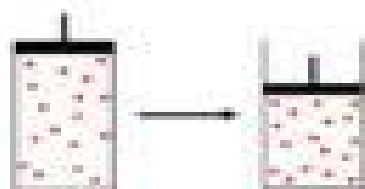
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Gas Law Practice

For the mathematical questions, show all of your work. For the conceptual questions, provide a thorough explanation for your answers.

[1] Convert 110°C into Kelvin.

A gas in a fixed cylinder as shown below is compressed while at a constant temperature. Initially, the pressure was exactly 1.00 atm, at a volume of 2.00 L. What is the final pressure when the cylinder is compressed to 1.40 L?



[2] A mass of neon occupies 500 mL at 100°C . Find its volume at 25°C , with the pressure remaining constant.

[3] A steel tank contains carbon dioxide at 27°C at a pressure of 6.0 atm. Determine the internal gas pressure when the tank and its contents are heated to 100°C .

Dalton Gas Law Practice Problems Key

Leo J. Malone, Theodore O. Dolter



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Roadmap to the Regents Sasha Alcott, 2003 If Students Need to Know It It is in This Book This book develops the chemistry skills of high school students It builds skills that will help them succeed in

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Chemical Principles with Qualitative Analysis William L. Masterton,Emil J. Slowinski,Conrad L. Stanitski,1986

Instructor's Solutions Manual to Accompany Atkins' Physical Chemistry, Ninth Edition C. A. Trapp,M. P. Cady,Carmen Giunta,2010 The Instructor s solutions manual to accompany Atkins Physical Chemistry provides detailed solutions to the b exercises and the even numbered discussion questions and problems that feature in the ninth edition of Atkins Physical Chemistry The manual is intended for instructors and consists of material that is not available to undergraduates The manual is free to all adopters of the main text

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