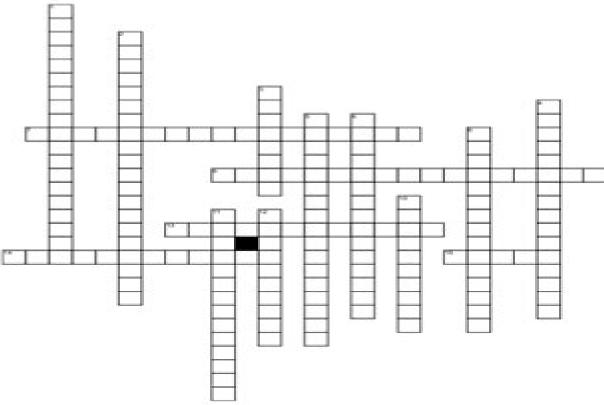
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Chemical Bonding



Across

bonding is a type of chemical bond where a pair of electrons is unequally shared between two atoms.

atoms share electrons through covalent bonds.

 a lasting attraction between atoms, ions or molecules that enables the formation of chemical compounds.

formed between two ions with opposite charges.

 a covalent bond between two atoms where the electrons forming the bond are unequally distributed.

Down

 electrons that are represented by dots; two dots between atoms represent a covalent bond.

two atoms share a pair of electrons with each other.

a group of atoms or 1 or 2 atoms or more

 a formula giving the number of atoms of each of the elements present in one molecule of a specific compound.

a formula that shows the arrangement of atoms in the molecule of a compound. A chemical bond formed by the sharing of one or more electrons, especially pairs of electrons, between atoms
 tells us the number of atoms

of each element in a compound.

10. energy required to break

 energy required to break apart a mole of molecules into its component atoms

 diagrams that show the bonding between atoms of a molecule

one pair of electrons is shared by two atoms in a molecule

Crossword Puzzles In Chemistry 6 Chemical Bonding Answers

Rose Arny

Crossword Puzzles In Chemistry 6 Chemical Bonding Answers:

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