

COVALENT BONDING

Name _____

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

electrons in the middle share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

wanted available share

<p>1. $\text{H} + \text{H} (\text{H}_2)$</p> <p>W - 4</p> <p>A - 2</p> <p>S - 2</p>	<p><i>wanted 2 + 2</i></p> <p><i>available</i></p> <p>$\text{H} \cdot \cdot \text{H}$</p> <p>$\text{H} \cdot \quad \cdot \text{H}$</p>	<p>$\text{H} - \text{H}$</p> <p><i>single bond = share 2 e⁻</i></p>
<p>2. $\text{F} + \text{F} (\text{F}_2)$</p> <p>W - 8 + 8 = 16</p> <p>A - 7 + 7 = 14</p> <p>S - 2</p>	<p>$\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot$</p> <p><i>2 e⁻</i></p>	<p>$\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot$</p> <p><i>single bond 2 e⁻</i></p>
<p>3. $\text{O} + \text{O} (\text{O}_2)$</p> <p>W - 8 + 8 = 16</p> <p>A - 6 + 6 = 12</p> <p><i>(S = 4)</i></p>	<p>$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot$</p> <p><i>4 e⁻</i></p>	<p>$\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot$</p> <p><i>4 double bond</i></p>
<p>4. $\text{N} + \text{N} (\text{N}_2)$</p> <p>W - 8 + 8 = 16</p> <p>A - 5 + 5 = 10</p> <p><i>S = 6</i></p>	<p>$\cdot \text{N} \cdot \cdot \cdot \text{N} \cdot$</p>	<p>$\cdot \text{N} \equiv \text{N} \cdot$</p>
<p>5. $\text{C} + \text{O} (\text{CO}_2)$</p> <p>W - 8 + 8 + 8 = 24</p> <p>A - 4 + 6 + 6 = 16</p> <p><i>S = 8</i></p>	<p>$\cdot \ddot{\text{O}} \cdot \cdot \cdot \text{C} \cdot \cdot \cdot \ddot{\text{O}} \cdot$</p>	<p>$\ddot{\text{O}} = \text{C} = \ddot{\text{O}}$</p>
<p>6. $\text{H} + \text{O} (\text{H}_2\text{O})$</p> <p>W - 2 + 2 + 8 = 12</p> <p>A - 1 + 1 + 6 = 8</p> <p><i>S = 4</i></p>	<p>$\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}$</p>	<p>$\text{H} - \ddot{\text{O}} - \text{H}$</p>

Covalent Bonding Section Assessment Answers

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Table of Contents Covalent Bonding Section Assessment Answers

1. Understanding the eBook Covalent Bonding Section Assessment Answers
 - The Rise of Digital Reading Covalent Bonding Section Assessment Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Bonding Section Assessment Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Bonding Section Assessment Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Bonding Section Assessment Answers
 - Personalized Recommendations
 - Covalent Bonding Section Assessment Answers User Reviews and Ratings

- Covalent Bonding Section Assessment Answers and Bestseller Lists
- 5. Accessing Covalent Bonding Section Assessment Answers Free and Paid eBooks
 - Covalent Bonding Section Assessment Answers Public Domain eBooks
 - Covalent Bonding Section Assessment Answers eBook Subscription Services
 - Covalent Bonding Section Assessment Answers Budget-Friendly Options
- 6. Navigating Covalent Bonding Section Assessment Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Covalent Bonding Section Assessment Answers Compatibility with Devices
 - Covalent Bonding Section Assessment Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covalent Bonding Section Assessment Answers
 - Highlighting and Note-Taking Covalent Bonding Section Assessment Answers
 - Interactive Elements Covalent Bonding Section Assessment Answers
- 8. Staying Engaged with Covalent Bonding Section Assessment Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covalent Bonding Section Assessment Answers
- 9. Balancing eBooks and Physical Books Covalent Bonding Section Assessment Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Bonding Section Assessment Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Covalent Bonding Section Assessment Answers
 - Setting Reading Goals Covalent Bonding Section Assessment Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Covalent Bonding Section Assessment Answers
 - Fact-Checking eBook Content of Covalent Bonding Section Assessment Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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