

# COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons  
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

*electrons in the middle share*

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

*wanted available share*

<p>1. <math>\text{H} + \text{H} (\text{H}_2)</math></p> <p>W - 4</p> <p>A - 2</p> <p>S - 2</p>	<p><i>wanted 2 + 2</i></p> <p><i>available</i></p> <p><math>\text{H} \cdot \cdot \text{H}</math></p> <p><math>\text{H} \cdot \quad \cdot \text{H}</math></p>	<p><math>\text{H} - \text{H}</math></p> <p><i>single bond = share 2 e<sup>-</sup></i></p>
<p>2. <math>\text{F} + \text{F} (\text{F}_2)</math></p> <p>W - 8 + 8 = 16</p> <p>A - 7 + 7 = 14</p> <p>S - 2</p>	<p><math>\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot</math></p> <p><i>2 e<sup>-</sup></i></p>	<p><math>\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot</math></p> <p><i>single bond 2 e<sup>-</sup></i></p>
<p>3. <math>\text{O} + \text{O} (\text{O}_2)</math></p> <p>W - 8 + 8 = 16</p> <p>A - 6 + 6 = 12</p> <p>S - 4</p>	<p><math>\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot</math></p> <p><i>4 e<sup>-</sup></i></p>	<p><math>\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot</math></p> <p><i>4 double bond</i></p>
<p>4. <math>\text{N} + \text{N} (\text{N}_2)</math></p> <p>W - 8 + 8 = 16</p> <p>A - 5 + 5 = 10</p> <p>S - 6</p>	<p><math>\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot</math></p>	<p><math>\cdot \ddot{\text{N}} \equiv \ddot{\text{N}} \cdot</math></p>
<p>5. <math>\text{C} + \text{O} (\text{CO}_2)</math></p> <p>W - 8 + 8 + 8 = 24</p> <p>A - 4 + 6 + 6 = 16</p> <p>S - 8</p>	<p><math>\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot</math></p>	<p><math>\cdot \ddot{\text{O}} = \text{C} = \ddot{\text{O}} \cdot</math></p>
<p>6. <math>\text{H} + \text{O} (\text{H}_2\text{O})</math></p> <p>W - 2 + 2 + 8 = 12</p> <p>A - 1 + 1 + 6 = 8</p> <p>S - 4</p>	<p><math>\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}</math></p>	<p><math>\text{H} - \ddot{\text{O}} - \text{H}</math></p>

# Covalent Bonding Gizmo Worksheet Answer Key

**Rosina Ehmann**



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## **Table of Contents Covalent Bonding Gizmo Worksheet Answer Key**

1. Understanding the eBook Covalent Bonding Gizmo Worksheet Answer Key
  - The Rise of Digital Reading Covalent Bonding Gizmo Worksheet Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Bonding Gizmo Worksheet Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Covalent Bonding Gizmo Worksheet Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Bonding Gizmo Worksheet Answer Key

- Personalized Recommendations
  - Covalent Bonding Gizmo Worksheet Answer Key User Reviews and Ratings
  - Covalent Bonding Gizmo Worksheet Answer Key and Bestseller Lists
5. Accessing Covalent Bonding Gizmo Worksheet Answer Key Free and Paid eBooks
- Covalent Bonding Gizmo Worksheet Answer Key Public Domain eBooks
  - Covalent Bonding Gizmo Worksheet Answer Key eBook Subscription Services
  - Covalent Bonding Gizmo Worksheet Answer Key Budget-Friendly Options
6. Navigating Covalent Bonding Gizmo Worksheet Answer Key eBook Formats
- ePub, PDF, MOBI, and More
  - Covalent Bonding Gizmo Worksheet Answer Key Compatibility with Devices
  - Covalent Bonding Gizmo Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Covalent Bonding Gizmo Worksheet Answer Key
  - Highlighting and Note-Taking Covalent Bonding Gizmo Worksheet Answer Key
  - Interactive Elements Covalent Bonding Gizmo Worksheet Answer Key
8. Staying Engaged with Covalent Bonding Gizmo Worksheet Answer Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Covalent Bonding Gizmo Worksheet Answer Key
9. Balancing eBooks and Physical Books Covalent Bonding Gizmo Worksheet Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Covalent Bonding Gizmo Worksheet Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Covalent Bonding Gizmo Worksheet Answer Key
- Setting Reading Goals Covalent Bonding Gizmo Worksheet Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Covalent Bonding Gizmo Worksheet Answer Key

- Fact-Checking eBook Content of Covalent Bonding Gizmo Worksheet Answer Key
- Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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