

Covalent Bonding Quiz (24 Points)**Section I – Basic Concepts & Vocabulary (12 points – 1 pt. each)**

Instructions: For Items 1–4, circle the correct response.

1. A *covalent bond* is formed by _____ electrons between two nonmetal atoms.
a) freezing c) sharing
b) transferring d) melting
2. A group of nonmetal atoms, joined by *covalent bonds* is defined as a(an)
a) molecule c) metal
b) salt crystal d) octet
3. A *structural formula* uses Lewis dot structures to show how atoms are bonded, including the number & types of bonds, in a covalent molecule? (Yes or No)
4. Do atoms, except hydrogen, in covalent bonding follow the octet rule? (Yes or No)

Instructions: Match the word on the left with the definition on the right by writing the letter on the line next to the word that is a “definition match” for that word.

- | | | | | |
|----------------------------|-------|--|-------|--|
| 5. molecular formula | _____ | A. a covalent bond in which only one pair of electrons is shared by two bonded atoms | | |
| | | 6. nonpolar covalent bond | _____ | B. chemical formula that shows the arrangement of atoms in molecules and polyatomic ions |
| 7. triple covalent bond | _____ | C. a covalent bond between two atoms of different electronegativities in which the bonding electrons are not shared equally | | |
| 8. double covalent bond | _____ | D. a covalent bond formed by the equal sharing of bonding electrons by two atoms | | |
| 9. polar covalent bond | _____ | E. a covalent bond in which two pairs of electrons are shared by two bonded atom | | |
| 10. single covalent bond | _____ | F. two valence electrons that are <u>not involved in bonding</u> at a bond site in a covalent molecule | | |
| 11. structural formula | _____ | G. indicates the exact number of atoms in a covalent compound, but does NOT show the arrangement of the atoms nor types of bonds | | |
| 12. unshared electron pair | _____ | H. a covalent bond in which three pairs of electrons are shared by two bonded atom | | |

Covalent Bonding Quiz Chemistry 3 Edgenuity

SJ Ball



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