

Name: Answer Key Date: _____ Per.: _____

Covalent Bond Practice

1. Fill in the missing information on the chart.

Element	# of Protons	# of Electrons	# of Valence Electrons	# of electrons to fill outer shell.
Carbon	6	6	4	4
Hydrogen	1	1	1	1
Chlorine	17	17	7	1
Helium	2	2	2	0
Phosphorus	15	15	5	3
Oxygen	8	8	6	2
Sulfur	16	16	6	2
Nitrogen	7	7	5	3

2. For each of the following covalent bonds:

- Write the symbols for each element.
- Draw a Lewis Dot structure for the valence shell of each element.
- Rearrange the electrons to pair up electrons from each atom.
- Draw circles to show the sharing of electrons between each pair of atoms.
- Draw the bond structure using chemical symbols and lines. Use one line for each pair of electrons that is shared.
- Write the chemical formula for each molecule.

a) Hydrogen + Hydrogen



b) Chlorine + Chlorine



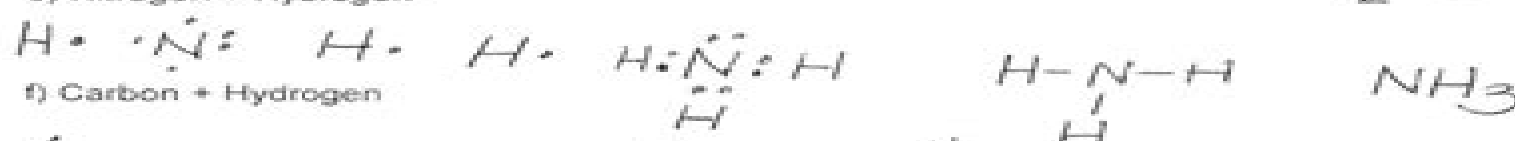
c) Hydrogen + Chlorine



d) Hydrogen + Oxygen



e) Nitrogen + Hydrogen



f) Carbon + Hydrogen



Covalent Bonding Practice Problems Answers

Arshad Iqbal



Covalent Bonding Practice Problems Answers:

Unveiling the Power of Verbal Art: An Psychological Sojourn through **Covalent Bonding Practice Problems Answers**

In a global inundated with screens and the cacophony of quick connection, the profound energy and psychological resonance of verbal beauty often disappear in to obscurity, eclipsed by the constant onslaught of noise and distractions. However, set within the musical pages of **Covalent Bonding Practice Problems Answers**, a interesting function of literary splendor that pulses with fresh thoughts, lies an remarkable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this magical opus manuals readers on a mental odyssey, softly exposing the latent potential and profound influence embedded within the delicate internet of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is key styles, dissect their charming writing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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