

Ionic and Covalent Naming Practice **Answers**

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

- | | | |
|-----|---------------------------|----------------------------------|
| 1) | Na_2CO_3 | sodium carbonate |
| 2) | P_4O_{10} | tetraphosphorus decaoxide |
| 3) | N_2O_5 | dinitrogen pentoxide |
| 4) | FeSO_4 | iron (II) sulfate |
| 5) | SiO_2 | silicon dioxide |
| 6) | FeCl_3 | iron (III) chloride |
| 7) | CoBr_2 | cobalt (II) bromide |
| 8) | B_2H_4 | diboron tetrahydride |
| 9) | CO | carbon monoxide |
| 10) | PCl_3 | phosphorus trichloride |

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

- | | | |
|-----|------------------------|--|
| 11) | dinitrogen trioxide | N_2O_3 |
| 12) | nitrogen dioxide | NO_2 |
| 13) | sulfur pentoxide | SO_5 |
| 14) | lithium acetate | $\text{LiC}_2\text{H}_3\text{O}_2$ |
| 15) | phosphorus trifluoride | PF_3 |
| 16) | vanadium (V) oxide | V_2O_5 |
| 17) | aluminum hydroxide | $\text{Al}(\text{OH})_3$ |
| 18) | zinc sulfide | ZnS |
| 19) | silicon tetrafluoride | SiF_4 |
| 20) | silver phosphate | Ag_3PO_4 |

Covalent Bonds Prentice Hall Teaching Resources

Answer Key

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Table of Contents Covalent Bonds Prentice Hall Teaching Resources Answer Key

1. Understanding the eBook Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - The Rise of Digital Reading Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Bonds Prentice Hall Teaching Resources Answer Key

- Personalized Recommendations
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key User Reviews and Ratings
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key and Bestseller Lists
5. Accessing Covalent Bonds Prentice Hall Teaching Resources Answer Key Free and Paid eBooks
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key Public Domain eBooks
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key eBook Subscription Services
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key Budget-Friendly Options
 6. Navigating Covalent Bonds Prentice Hall Teaching Resources Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key Compatibility with Devices
 - Covalent Bonds Prentice Hall Teaching Resources Answer Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Highlighting and Note-Taking Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Interactive Elements Covalent Bonds Prentice Hall Teaching Resources Answer Key
 8. Staying Engaged with Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covalent Bonds Prentice Hall Teaching Resources Answer Key
 9. Balancing eBooks and Physical Books Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Bonds Prentice Hall Teaching Resources Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Setting Reading Goals Covalent Bonds Prentice Hall Teaching Resources Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Covalent Bonds Prentice Hall Teaching Resources Answer Key

- Fact-Checking eBook Content of Covalent Bonds Prentice Hall Teaching Resources Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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