

Name: \_\_\_\_\_

Sci # \_\_\_\_\_

## COVALENT BOND WEBQUEST

DIRECTIONS: Answer the following by going to the notes website. You can go to Mrs. Slowik's website and use the links or type in the addresses for each section if needed.

### Activity #1- Introduction to Covalent Bonding

Open [Chemical Bonding](#). Scroll down to the heading "Covalent Bonding." OR  
[http://www.visionlearning.com/library/module\\_viewer.php?mid=55](http://www.visionlearning.com/library/module_viewer.php?mid=55)

1. As opposed to \_\_\_\_\_ bonding in which a complete transfer of electrons occur, \_\_\_\_\_ bonding occurs when two or more elements \_\_\_\_\_ electrons. Covalent bonding occurs because the atoms in the compound have a similar tendency for electrons (generally to \_\_\_\_\_ electrons). This most commonly occurs when two \_\_\_\_\_ bond together. Because both of the nonmetals will want to \_\_\_\_\_ electrons, the elements involved will share electrons in an effort to \_\_\_\_\_ their valence shells.

2. How many valence electrons are in one atom of hydrogen?

3. How many valence electrons does hydrogen need to have a full first shell?

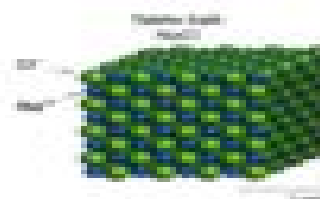
4. How does the hydrogen atom "pick up" another electron?

5. What compound does hydrogen form?

6. How do hydrogen atoms make a covalent bond?

7. Visit the simulation, [Covalent bonding between hydrogen atoms](#). OR  
[http://www.visionlearning.com/library/flash\\_viewer.php?fid=1346&mid=55](http://www.visionlearning.com/library/flash_viewer.php?fid=1346&mid=55)  
Describe or draw what you see.

8. Label the substances below as "ionic" or "covalent".



# Covalent Bonding Webquest Answers Key

**Margaret Irene Lindsay**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

## Covalent Bonding Webquest Answers Key:

### Covalent Bond 35 Success Secrets - 35 Most Asked Questions on Covalent Bond - What You Need to Know

Shirley Soto,2014-10-14 A Blue Ribbon Covalent bond Guide A covalent bond is a biochemical bond that includes the parting of negatron matches amid particles The steady level of alluring and hideous drives amid particles once they share electrons is recognized like covalent joining For numerous particles the parting of electrons permits every one particle to attain the equal of a complete outside shell comparable to a steady microelectronic arrangement There has never been a Covalent bond Guide like this It contains 35 answers much more than you can imagine comprehensive answers and extensive details and references with insights that have never before been offered in print Get the information you need fast This all embracing guide offers a thorough view of key knowledge and detailed insight This Guide introduces what you want to know about Covalent bond A quick look inside of some of the subjects covered Chemical bonding Covalent bond Noncovalent bonding Drug Design Covalent bond History Noncovalent bonding Cation Anion Noncovalent bonding Hydrophobic effect Polar covalent bond Polarity of bonds Noncovalent bonding Boiling Points of Liquids Noncovalent bonding London Dispersion Forces Coordinate Covalent Bond Examples Noncovalent bonding effects Noncovalent bonding H bonding Noncovalent bonding Polar Polar covalent bond Polar molecules Covalent bonds Polarity of covalent bonds Polar covalent bond Polarity of molecules Noncovalent bonding Interaction Noncovalent bonding Van der Waals Forces Covalent bonds Subdivision of covalent bonds Covalent Bond Classification Polar covalent bond Hybrids Noncovalent bonding Electrostatic Interactions Polar covalent bond Nonpolar molecules Covalent Bond Classification Other uses Covalent bonds History Noncovalent bonding Dipole Dipole Noncovalent bonding Protein Folding Structure and much more *Covalent Bonding* ,2006 **The Covalent Bond** Henry Sinclair Pickering,1977 *Covalent Bonding in Crystals, Molecules, and Polymers* James C. Phillips,1969 **Chemistry of the Covalent Bond** Leallyn Burr Clapp,1957 Chemical Bonding Kok Koon Leong,2021-05-03 This book introduces the principles behind chemical bonding to teenagers between the ages of fifteen to seventeen Topics covered include ionic bonding covalent bonding and metallic bonding **Local Reversibility and the Calculus of Covalent Bonding** Stefan Kuhn,2019 **Teaching Chemical Bonding** Margaret Irene Lindsay,1995 This document presents an instructional strategy for teaching chemical bonding using parables and music Games student interactions and worksheets are included in the lesson plans Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding JRH The Study of Covalent-Bonding by Means of Elastic-Constants ,1973

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