

Name: \_\_\_\_\_

Sci # \_\_\_\_\_

## COVALENT BOND WEBQUEST

DIRECTIONS: Answer the following by going to the notes website. You can go to Mrs. Slowik's website and use the links or type in the addresses for each section if needed.

### Activity #1- Introduction to Covalent Bonding

Open [Chemical Bonding](#). Scroll down to the heading "Covalent Bonding." OR  
[http://www.visionlearning.com/library/module\\_viewer.php?mid=55](http://www.visionlearning.com/library/module_viewer.php?mid=55)

1. As opposed to \_\_\_\_\_ bonding in which a complete transfer of electrons occur, \_\_\_\_\_ bonding occurs when two or more elements \_\_\_\_\_ electrons. Covalent bonding occurs because the atoms in the compound have a similar tendency for electrons (generally to \_\_\_\_\_ electrons). This most commonly occurs when two \_\_\_\_\_ bond together. Because both of the nonmetals will want to \_\_\_\_\_ electrons, the elements involved will share electrons in an effort to \_\_\_\_\_ their valence shells.

2. How many valence electrons are in one atom of hydrogen?

3. How many valence electrons does hydrogen need to have a full first shell?

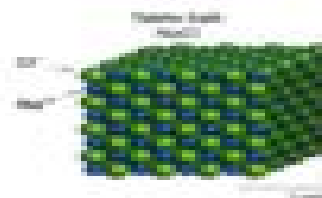
4. How does the hydrogen atom "pick up" another electron?

5. What compound does hydrogen form?

6. How do hydrogen atoms make a covalent bond?

7. Visit the simulation, [Covalent bonding between hydrogen atoms](#). OR  
[http://www.visionlearning.com/library/flash\\_viewer.php?mid=124&mid=55](http://www.visionlearning.com/library/flash_viewer.php?mid=124&mid=55)  
Describe or draw what you see.

8. Label the substances below as "ionic" or "covalent".



# Covalent Bond Webquest Key

**D Kirk**



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