

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
crisscross

noble gas
electrons
anion

octet
ions
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.

10. A cation has a positive charge.

11. A binary ionic compound contains only one kind of cation and one kind of anion.

12. Monatomic ions consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.

14. A great deal of energy is produced when an ionic compound is formed.

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RM Cervero



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Chemical Engineering Farooq Sher,2025-10-04 Artificial Intelligence in Chemical Engineering explores the integration of artificial intelligence AI into various facets of chemical engineering The book introduces historical information highlights current state and trends in AI applications and discusses challenges and opportunities within the field Foundational principles of AI and machine learning are thoroughly covered giving readers a solid understanding of basic AI principles machine learning algorithms and the crucial processes of model training and validation The book then delves into the critical phase of data acquisition and preprocessing for AI models addressing strategies for data collection ensuring data quality and techniques for feature engineering and selection Subsequent chapters cover a wide spectrum of AI applications in chemical engineering From supervised and unsupervised learning for process modeling to the advanced realm of deep learning applications this book explores neural networks convolutional and recurrent architectures and their real world applications in process optimization and analysis Navigates the dynamic intersection of AI and chemical engineering covering ethical considerations interdisciplinary applications and AI s impact on safety sustainability and innovation Bridges the gap between policy and implementation of AI in chemical engineering facilitating a harmonious integration of AI technologies and fostering responsible and effective use within the chemical engineering industry Offers a forward looking approach to guide professionals researchers and students in navigating the dynamic and transformative future of AI in chemical engineering

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