

**Chemistry 101 – General Chemistry
Practice Final Exam**

Name _____

h = 6.626×10^{-34} J·s (Planck's constant)

c = 3.00×10^8 m/s (speed of light)

$R_g = 8.314$ J/mol·K (gas constant)

Strategies of Solvers (8 points each)

Identify the name of the value that best completes the statement or answers the question and highlight (circle) the letter in the space preceding the problem number.

- _____ 1. Which of the following is not a colligative?
a. molality
b. molarity
c. mass
d. ΔT
e. freezing point
- _____ 2. Which of the following statements concerning the photo effect?
a. Increased wavelength yields the electrons.
b. The photo current increases, increases with frequency.
c. Frequency and wavelength affect the reaction.
d. The reaction concerns both photons and electrons.
e. Most of the mass of the photo is found in the nucleus.
- _____ 3. Which of the following statements about two isotopes is false?
a. They will have the same charge on the nucleus.
b. They will have different numbers of neutrons.
c. They will have essentially the same chemical reactivity.
d. They will have the same atomic number.
e. They will have the same atomic weight.
- _____ 4. Which element can be classified as an effective p-block?
a. Al
b. Mg
c. Fe
d. Sn
e. Br
- _____ 5. Which combination below corresponds to reaction with a +1 charge?
a. 7 protons and 8 electrons
b. 11 protons and 10 electrons
c. 10 protons and 11 electrons
d. 11 protons and 11 electrons
e. 10 protons and 11 electrons

Csula Chemistry Final Exam

David E. Goldberg



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The Hidden Curriculum—Faculty-Made Tests in Science Sheila Tobias, Jacqueline Raphael, 2013-06-29 This resource manual for college level science instructors reevaluates the role of testing in their curricula and describes innovative techniques pioneered by other teachers part I examines the effects of the following on lower division courses changes in exam content format and environment revisions in grading practices student response colleague reaction the sharing of new practices with other interested professionals and more The book includes a comprehensive introduction faculty composed narratives commentaries by well known science educators and a visual index to 100 more refined innovations **Human Exceptionality** Michael L. Hardman, 1993 [Life on the Dunes](#) Brian M. Fagan, 2006 Highlights archeological findings on the remote San Nicolas Island off the coast of the California Includes tables graphs and photographs **Abbreviations Dictionary** Ralph De Sola, 1981 Dictionary of abbreviations etc **Research Resources Reporter**, 1986 *Cal-OSHA Reporter*, 1981 [AGU 2004 Joint Assembly](#) American Geophysical Union. Joint Assembly, 2004 **Who's who in California**, 1990 [Profiles of American Colleges -- 2008](#) Barron's Educational Series, 2008-07-01 Up to date facts and figures on enrollments tuition and fees academic programs campus environment available financial aid and much more make the 28th edition of Profiles of American Colleges America's most authoritative data source for college bound high school students their parents and high school guidance counselors More than 1 650 accredited four year colleges are profiled An interactive CD ROM enclosed with the directory guides students to specific schools when they enter details describing their personal academic plans and aptitudes In addition to the above cited information each college profile gives details on Admission requirements Library and computer facilities Admissions procedures for freshmen Campus safety and security Thumbnail descriptions of faculty Requirements for a degree Athletic facilities Extracurricular activities E mail addresses College fax numbers and web sites Admissions Contacts and much more Schools are rated according to Barron's well known competitiveness scale from Noncompetitive to Most Competitive Unlike some other publications Barron's refrains from the unreliable practice of ranking colleges on a first through last basis The book's tinted pages section presents a quick reference Index of College Majors that lists all available major study programs at each school Also profiled are many excellent colleges in Canada and several other countries as well as brief profiles of religious colleges and American colleges based in foreign countries **CHEM 1792**, 2012 **Chemistry** John Coggins, 2005 *Chemistry* John Coggins, 2001 **McGraw-Hill's 500 College Chemistry Questions** David E. Goldberg, 2012-09-07 500 Ways to Achieve Your Best Grades We want you to succeed on your college chemistry midterm and final exams That's why we've selected these 500 questions to help you study more effectively use your preparation time wisely and get your best grades These questions are similar to the ones you'll find on a typical college exam so you will know what to expect on test day Each question includes comprehensive explanations in the answer key Whether you have been studying all year or are doing a last minute review

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