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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain **electrons**. The outermost electrons are called **valence electrons**.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

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S Ashworth



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Table of Contents Covealent Bonds Student Exploration Gizmo Worksheet Answers

1. Understanding the eBook Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - The Rise of Digital Reading Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Personalized Recommendations
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers User Reviews and Ratings
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers and Bestseller Lists

5. Accessing Covealent Bonds Student Exploration Gizmo Worksheet Answers Free and Paid eBooks
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers Public Domain eBooks
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers eBook Subscription Services
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers Budget-Friendly Options
6. Navigating Covealent Bonds Student Exploration Gizmo Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers Compatibility with Devices
 - Covealent Bonds Student Exploration Gizmo Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Highlighting and Note-Taking Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Interactive Elements Covealent Bonds Student Exploration Gizmo Worksheet Answers
8. Staying Engaged with Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covealent Bonds Student Exploration Gizmo Worksheet Answers
9. Balancing eBooks and Physical Books Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covealent Bonds Student Exploration Gizmo Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Setting Reading Goals Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Fact-Checking eBook Content of Covealent Bonds Student Exploration Gizmo Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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