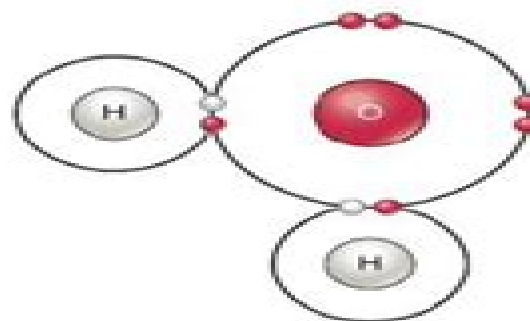
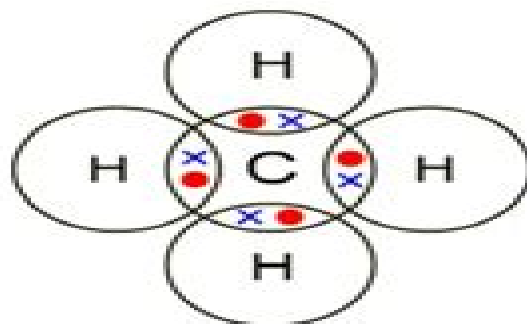
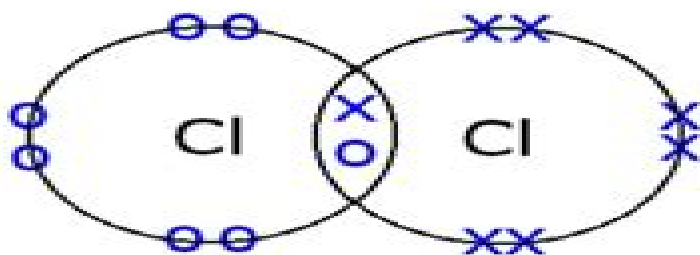


COVALENT BONDING

- Chemical bond formed by sharing of electrons (from the valence shell) between two atoms, so that each atom in the resulting molecule gets the stable electronic configuration of the nearest noble gas.
- A covalent bond is formed only when both reacting atoms need electrons to achieve stable electronic configuration. ie, Non-metal atoms form covalent bonds
- When two non-metal atoms come close enough, their shells overlap and they share electrons. The strong force of attraction developed between the atoms when they share electrons is called covalent bond.
- Three types of covalent bond – Single , Double , Triple – based on the number of electrons shared by each non-metal atom pair.



Covalent Bond Notes Ch 8

R D Madan



Covalent Bond Notes Ch 8 :

Unveiling the Magic of Words: A Review of "**Covalent Bond Notes Ch 8** "

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Covalent Bond Notes Ch 8** ," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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