Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Frank can share his red marker

with Flora and Flora can share her blue marker with frank.

Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? Olivia can share her purple and green markers with otta and otta can share his black and brown markers with olivia.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covarent Bond's Gizmo, atoms form bonds in this way.





- The outermost electrons in each atom are called valence electrons. How many valence electrons does each fluorine atom have? Each fluorine atom has seven valence electrons.
- Click Pause (!=). Drag an electron from the left atom to the right atom. Click Play. What happens? The electrons that was dragged now orbits both atoms.
- Click Pause, drag an electron from the right atom to the left, and then click Play. What happens now? A pair of electrons now orbit both atoms.

Covalent Bond Explore Learning Answer Key

Maria C. Grant, Douglas Fisher

Covalent Bond Explore Learning Answer Key:

Chemistry Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives Psychology of Learning Science Shawn M. Glynn, Bruce K. Britton, Russell H. Yeany, 2012-11-12 Focusing on the teaching and learning of science concepts at the elementary and high school levels this volume bridges the gap between state of the art research and classroom practice in science education The contributors science educators cognitive scientists and psychologists draw clear connections between theory research and instructional application with the ultimate goal of improving science teachers effectiveness in the classroom Toward this end explicit models illustrations and examples drawn from actual science classes are included **Learning with Understanding in the Chemistry Classroom** Iztok Devetak, Saša Aleksij Glažar, 2014-01-14 This volume offers a critical examination of a variety of conceptual approaches to teaching and learning chemistry in the school classroom Presenting up to date research and theory and featuring contributions by respected academics on several continents it explores ways of making knowledge meaningful and relevant to students as well as strategies for effectively communicating the core concepts essential for developing a robust understanding of the subject Structured in three sections the contents deal first with teaching and learning chemistry discussing general issues and pedagogical strategies using macro sub micro and symbolic representations of chemical concepts Researchers also describe new and productive teaching strategies. The second section examines specific approaches that foster learning with understanding focusing on techniques such as cooperative learning presentations laboratory activities multimedia simulations and role playing in forensic chemistry classes The final part of the book details learner centered active chemistry learning methods active computer aided learning and trainee chemistry teachers use of student centered learning during their pre service education Comprehensive and highly relevant this new publication makes a significant contribution to the continuing task of making chemistry classes engaging and effective Teaching and Learning STEM Richard M. Felder, Rebecca Brent, 2024-03-13 The widely used STEM education book updated Teaching and Learning STEM A Practical Guide covers teaching and learning issues unique to teaching in the science technology engineering and math STEM disciplines Secondary and postsecondary instructors in STEM areas need to master specific skills such as teaching problem solving which are not regularly addressed in other teaching and learning books This book fills the gap addressing topics like learning objectives course design choosing a text effective instruction active learning teaching with technology and assessment all from a STEM perspective You ll also gain the knowledge to implement learner centered instruction which has been shown to improve learning outcomes across disciplines For this edition chapters have been updated to reflect recent cognitive science and empirical educational research findings that inform STEM pedagogy You ll

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