# 12

# STOICHIOMETRY

# **Practice Problems**

In your notations, pales the following problems.

#### SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Ohe the 3-copy problem, soluting approach you featured to Obspary 1.

- An apple pin needs 10 large apples, 2 streets (top and bottom), and 1 tablespeen of closuston. Write a balanced equation that fits this situation. How many apples are needed to make 26 piles?
- Two modes of points states ablieded and three modes of stagged are produced from the decomposition of two males of patentiam obligates, 8020(5):1. Write the finitesized equation. How many makes of oxygen are produced from 12 modes of points at the control.
- Using the equation from problem 2, how many makes of oxygen are produced from 14 reales of potassium chlomatic
- Two modecules of hydrogen resor with one molecule of oxyges to produce two molecules of water. How many molecules of water are produced from 2.8 × 10<sup>24</sup> molecules of oxygen? How many males of water are produced from 22.5 nodes of oxygen?

#### SECTION 12.2 CHEMICAL CALCULATIONS

 Calculate the number of males of hydrogen chloride produced from 10 moles of featnesses.

 Calculate the number of states of objective meeted to form 14 moles of insu(33) objectes.

 Culculate the number of grams of strongers disolds that are produced from a moles of sitric oxide.

$$2590 \log + O_{\rm phy} \rightarrow 2590 \rm phy$$

 Calculate the mass of orgges produced from the decomposition of PLO g of posterium chlorum.

- B. Calculate the mass of eliver needed to must with objection in produce 64 g of silver objected. (What Write a leakeneed equation first):
- How entity liters of authors momentale at STP are specied to return with 4.00 g; of oxygen gas to produce authors clioside?

 Calculate the number of lines of organ gas needed to produce 15.0 liters of distingen trioxide. Assume all gases are at the same conditions of temperature and pressure.

$$3N_{\rm e}kp = 30_{\rm e}kp \rightarrow 2N_{\rm e}0_{\rm e}kp$$

# **Core Teaching Resources Chemistry Answers Chp 13**

**Michael Rajnik** 

# **Core Teaching Resources Chemistry Answers Chp 13:**

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