



Equations

Name _____

Date _____

Write a balanced chemical equation to represent each of the following chemical reactions.

1. iron + sulfur \rightarrow iron(II) sulfide
2. zinc + copper(II) sulfate \rightarrow zinc sulfate + copper
3. silver nitrate + sodium bromide \rightarrow sodium nitrate + silver bromide
4. potassium chlorate (heated) \rightarrow potassium chloride + oxygen
5. water (electricity) \rightarrow hydrogen + oxygen
6. mercury(II) oxide (heated) \rightarrow mercury + oxygen
7. potassium iodide + lead(II) nitrate \rightarrow lead(II) iodide + potassium nitrate
8. aluminum + oxygen \rightarrow aluminum oxide
9. magnesium chloride + ammonium nitrate \rightarrow magnesium nitrate + ammonium chloride
10. iron(III) chloride + ammonium hydroxide \rightarrow iron(III) hydroxide + ammonium chloride
11. sodium peroxide + water \rightarrow sodium hydroxide + oxygen
12. iron(III) oxide + carbon \rightarrow iron + carbon monoxide
13. iron + water \rightarrow hydrogen + iron(III) oxide
14. iron(III) chloride + potassium hydroxide \rightarrow potassium chloride + iron(III) hydroxide
15. aluminum + sulfuric acid \rightarrow aluminum sulfate + hydrogen
16. sodium carbonate + calcium hydroxide \rightarrow sodium hydroxide + calcium carbonate



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M Tight



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