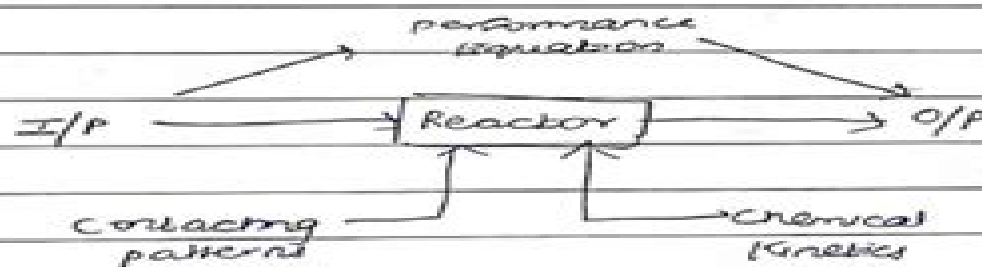


- 1) Kinetics of Homogeneous Reactions
- 2) Conversion and reactor sizing
- 3) Analysis of Rate data
- 4) Isothermal Reactor Design
- 5) Non-Isothermal Reactor Design
- 6) Analysis of non-Ideal ~~Reactors~~ reactors
- 7) RTD for chemical ~~reactors~~ reactors (Residence Time Distribution)



Performance equation = $f(\text{I/P, Contacting pattern, Chemical Kinetics})$

Classification of Reactions

- Based on phases - Homogeneous $\rightarrow NO_2(g) + \frac{1}{2} O_2(g) \rightarrow NO_2(g)$
 Heterogeneous $\rightarrow C_2H_5OH(l) + CH_3COOH(l) \xrightarrow{H_2SO_4} CH_3COOC_2H_5(l) + H_2O(l)$
 $SO_2(g) + \frac{1}{2} O_2(g) \xrightarrow{V_2O_5} SO_3(g)$
 But catalyst is solid phase

- 3) Based on catalytic or non-catalytic
- 4) Based on no. of molecules that are taking part in the reaction

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ML Morrison



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