

SCH 2202 ORGANIC CHEMISTRY II

► JANUARY-APRIL 2024

► JKUAT- DEPT OF CHEMISTRY

BENZENE AND AROMATICITY

- **Benzene** (C_6H_6) is the simplest **aromatic hydrocarbon** (or arene).
- Four degrees of unsaturation.
- It is planar.
- All C—C bond lengths are equal.
- Whereas unsaturated hydrocarbons such as alkenes, alkynes and dienes readily undergo addition reactions, benzene does not.



- Benzene reacts with bromine only in the presence of $FeBr_3$ (a Lewis acid), and the reaction is a substitution, not an addition.



STRUCTURE OF BENZENE

- The true structure of benzene is a resonance hybrid of the two Lewis structures.

The hybrid

Some texts draw benzene as a hexagon with an inner circle:

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