



Chemistry

Vocabulary Puzzle



Covalent Bonding

Write the following vocabulary words and definitions in your own words.

1. Covalent Bond - The force that holds two atoms together by sharing a pair of electrons.
2. Substitution Reaction - A chemical reaction in which one atom or group of atoms in a molecule is replaced by another atom or group of atoms.
3. Double Bond - A chemical bond between two atoms consisting of two covalent bonds.
4. Lewis Structure - A diagram that shows the arrangement of valence electrons in a molecule.
5. Molecule - A group of two or more atoms held together by covalent bonds.
6. Single Bond - A bond that is formed when two atoms share one pair of electrons.
7. Triple Bond - A single covalent bond that is formed when two atoms share three pairs of electrons.
8. Compound - A substance that contains two or more different elements.
9. Coordinate Covalent Bond - A bond in which one atom provides both electrons for the bond.
10. Hydrocarbon - A compound that contains only carbon and hydrogen atoms.
11. Structural Formula - A diagram that shows the relative positions of atoms in a molecule.
12. Hybridization - A process in which atomic orbitals combine to form hybrid orbitals.
13. Hybrid Orbital - An orbital that is formed by the combination of atomic orbitals.
14. Polar Covalent Bond - A type of covalent bond in which the electrons are shared unequally between two atoms.
15. Nonpolar Covalent Bond - A type of covalent bond in which the electrons are shared equally between two atoms.

Name: _____

Covalent

Answer:

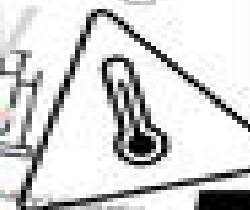
1. A chemical reaction in which one atom or group of atoms in a molecule is replaced by another atom or group of atoms.
2. A chemical bond between two atoms consisting of two covalent bonds.
3. A diagram that shows the arrangement of valence electrons in a molecule.
4. A group of two or more atoms held together by covalent bonds.
5. A bond that is formed when two atoms share one pair of electrons.
6. A single covalent bond that is formed when two atoms share three pairs of electrons.
7. A compound that contains two or more different elements.
8. A bond in which one atom provides both electrons for the bond.
9. A diagram that shows the relative positions of atoms in a molecule.
10. A process in which atomic orbitals combine to form hybrid orbitals.
11. An orbital that is formed by the combination of atomic orbitals.
12. A type of covalent bond in which the electrons are shared unequally between two atoms.
13. A type of covalent bond in which the electrons are shared equally between two atoms.
14. A bond that is formed when two atoms share one pair of electrons.
15. A diagram that shows the relative positions of atoms in a molecule.

Draw:

1. A molecule in which the atoms are covalently bonded.
2. A bond between two atoms where the electronegativity difference is less than 0.4.
3. The force that holds two atoms together by sharing a pair of electrons.
4. A type of bond that is formed when two atoms share one pair of electrons.
5. A bond that is formed when two atoms share three pairs of electrons.
6. The ability of an atom to attract electrons.
7. A condition that exists when the electrons in a bond are shared unequally.
8. A molecular orbital that is formed by the combination of atomic orbitals.
9. A model that uses electron-dot diagrams to represent molecular structure.
10. Forces that hold two atoms together by sharing a pair of electrons.

Name: _____

Covalent Bonding



Covalent Bonding

Core Teaching Resources Chemistry Covalent Bonding

Vocabulary Review

ML Yell

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