

Naming and Writing Formulas for Covalent (Molecular) Compounds

Name: _____

Part A: Name the covalent compounds.

1. CO carbon monoxide
2. CO_2 carbon dioxide
3. N_2O_3 dinitrogen trioxide
4. NP nitrogen monophosphide
5. SCl_2 sulfur dichloride
6. P_2O_5 diphosphorus pentoxide
7. NBr_3 nitrogen tribromide
8. Cl_4 chlorine tetrachloride
9. CCl_4 carbon tetrachloride
10. PF_5 phosphorus pentafluoride
11. SeF_2 selenium difluoride
12. TeBr_2 tellurium dibromide
13. P_2S_5 diphosphorus pentasulfide
14. C_3N_4 tricarbon tetranitrogen
15. CH_4 carbon tetrahydride
16. PH_3 phosphorus trihydride

Part B: Write the Chemical Formula

1. carbon tetrafluoride CF_4
2. silicon dioxide SiO_2
3. dinitrogen trisulfide N_2S_3
4. phosphorus mononitride PN
5. carbon disulfide CS_2
6. dinitrogen trichloride N_2Cl_3
7. silicon tetrabromide SiBr_4
8. carbon dioxide CO_2
9. nitrogen trifluoride NF_3
10. boron trisulfide BS_3
11. disulfur trioxide S_2O_3
12. selenium tetrafluoride SeF_4
13. diphosphorus pentasulfide P_2S_5
14. sulfur dibromide SBr_2
15. carbon tetrachloride CCl_4
16. dinitrogen tetroxide N_2O_4

Building And Naming Molecules Lab Notes Answer Key

Sebastian Brünink



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