Chemical Bonding Review

n e	em	nical Bonding Re	viev	Name:	
ext	ions	: Answer each of the following	ng ques	stions by circling the correct answer choice.	
1.	340	hich two elements have the s	ame nu	imber of valence electrons?	
	A	Boron and Aluminum	C.	Argon and Helium	
	В.	Carbon and Nitrogen	D.	. Silicon and Aluminum	
2.	WW	What type of bond forms when electrons are transferred?			
	A	Covalent	C.	Metallic	
	8.	Ionic	D.	Nonmetallic	
3.	What is the chemical symbol for a potassium ion?				
-		K-1		P*1	
	6.	K.,	D.	. pr.1	
4.			-	. pr ³ filled outermost energy level in the Lewis-dot structure below	
4.			d for a f	filled outermost energy level in the Lewis-dot structure below	
4	Ho	• Ç •	d for a f	filled outermost energy level in the Lewis-dot structure below	
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	A B.	w many electrons are needed 4 0 hen atoms gain electrons duri	C. D. ing cher	filled outermost energy level in the Lewis-dot structure below 2 1 mical bonding, they become Negatively charged ions	
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5.	A B. WI A B.	w many electrons are needed 4 0 hen atoms gain electrons duri	C. D. ing cher C.	filled outermost energy level in the Lewis-dot structure below 2 1 mical bonding, they become Negatively charged ions Chemically charged ions	
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Bond Test Review Key Answers

Ray Holder

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