Name

Bond energy is defined as the amount of energy required to break a bond. These values are positive, indicating that bond breaking is endothermic. Bond energies are reported in kilojoules per mole (kJ/mol). The energy for breaking a hydrogen-hydrogen bond is 436 kJ/mol so when a hydrogen-hydrogen bond is formed the process releases 436 kJ/mol...

In a chemical reaction several bonds are broken and formed. For example in the reaction below a hydrogen-hydrogen bond is broken and a fluorine-fluorine bond is broken. Two hydrogen-fluorine bonds are formed. The overall energy change for this process is calculated below.

H-HBland energy: 436 kJ/mol

F-FBond Energy: 155 kJ/mol

example
$$H_2(g) + F_2(g) \rightarrow 2HF$$

$$\Delta H_m = [\text{energy used for breaking bonds}] - [\text{energy formed in making bonds}]$$

$$[436 \text{ kJ/mol} + 155 \text{ kJ/mol}] - [2(567 \text{ kJ/mol})]$$

$$= -543 \text{ kJ/mol}$$

Average Bond Energies (kJ/mol)

| H-H | 4.36 kJ/mod | C-H | 413 kl/mol | C=C | 614 kJ//mol |
|------|--------------|-------|--------------|-------|-------------|
| H-C1 | 431 kJ/mol | C-C | 348 kil/mol | CHC | 839 kJVmol |
| H-F | 567 kJ/mol | C-N | 293 kJ/mol | C=O | 799 kJ//mol |
| N-H | 3/91 kJ/molt | C-O | 3.58 kJVmod | O=O | 495 kJ//mol |
| N-O | 201 kJ/mol | C-F | 485 kJ/mod | CHO | 1072 kJ/mod |
| O-H | 463 k3/mol | C-CI | 328 kJ/mol | CHN | 615 k.Wmell |
| O-O | 146 kJ/molt | C-S | 2.59 kJ//mol | N-N | 418 kJVmol |
| F-F | 155 kJ/mol | CI-CI | 242 kJ/mol | NewNo | 941 kJ/mol |
| | | | | CHN | 891 kJ/mol |

Estimate the enthalpy change (ΔH_{res}) of the following reactions using the bond energies above.

1.
$$H-H+CI-CI \rightarrow H-CI+H-CI$$
4. $H-\frac{CI}{C}-CI+O \rightarrow \frac{O}{C}-CI+H-CI$
2. $H-C=C + H-F-F \rightarrow H-\frac{H}{C}-\frac{H}{C}-H$
5. $H-\frac{C}{C}-H+O=O \rightarrow O=C=O+H-O-H$
3. $H-\frac{H}{C}-O-H+H-\frac{H}{C}-O-H \rightarrow H-\frac{H}{C}-O-H+H-O-H$

Draw Lewis structures for reactants and products. Estimate the enthalpy change (ΔH_{re}) for the reactions using bond energies.

9.
$$N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$$

10.
$$H_2(g) + C_2H_4(g) \rightarrow C_2H_6(g)$$

11.
$$2C_2H_6(g) + 7O_2(g) \rightarrow 4CO_2(g) + 6H_2O(g)$$

13. HCN
$$(g)$$
 + 2H₂ (g) \rightarrow CH₃NH₂ (g)

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Chemistry Extension File Eileen Ramsden,2002 This chemistry extension file includes teaching notes guidance on coursework activities and equipment It has at least one assignment for each topic in the textbooks suitable for classwork and homework A comprehensive range of practical activities are included It contains extensive Key Skills and ICT materials An exam file resource containing a complete set of exam style questions in a format that can be used throughout Years 10 and 11 or as a resource for a revision programme is included

Objective Question Bank in Chemistry B.K. Sharma,1997

Measurement Technology for Micro-Nanometer Devices Wendong Zhang, Xiujian Chou, Tielin Shi, Zongmin Ma, Haifei Bao, Jingdong Chen, Liquo Chen, Dachao Li, Chenyang Xue, 2017-01-17 A fully comprehensive examination of state of the art technologies for measurement at the small scale Highlights the advanced research work from industry and academia in micro nano devices test technology Written at both introductory and advanced levels provides the fundamentals and theories Focuses on the measurement techniques for characterizing MEMS NEMS devices A Level Chemistry for OCR A: Year 1 and AS Rob Ritchie, Dave Gent, 2016-05-05 Please note this title is suitable for any student studying Exam Board OCR Level A Level Year 1 and AS Subject Chemistry First teaching September 2015 First exams June 2016 Written by curriculum and specification experts this Student Book supports and extends students throughout their course whilst delivering the breadth depth and skills needed to succeed at A Level and beyond Computational Chemistry Errol G. Lewars, 2010-11-10 This corrected second edition contains new material which includes solvent effects the treatment of singlet diradicals and the fundamentals of computational chemistry Computational Chemistry Introduction to the Theory and Applications of Molecular and Quantum Mechanics is an invaluable tool for teaching and researchers alike The book provides an overview of the field explains the basic underlying theory at a meaningful level that is not beyond beginners and it gives numerous comparisons of different methods with one another and with experiment The following concepts are illustrated and their possibilities and limitations are given potential energy surfaces simple and extended Hueckel methods ab initio AM1 and related semiempirical methods density functional theory DFT Topics are placed in a historical context adding interest to them and removing much of their apparently arbitrary aspect The large number of references to all significant topics mentioned should make this book useful not only to undergraduates but also to graduate students and academic and industrial researchers

Reliability Physics and Engineering J. W. McPherson,2018-12-20 This third edition textbook provides the basics of reliability physics and engineering that are needed by electrical engineers mechanical engineers civil engineers biomedical engineers materials scientists and applied physicists to help them to build better devices products The information contained within should help all fields of engineering to develop better methodologies for more reliable product designs more reliable materials selections and more reliable manufacturing processes all of which should help to improve product reliability A mathematics level through differential equations is needed Also a familiarity with the use of excel spreadsheets is assumed

Any needed statistical training and tools are contained within the text While device failure is a statistical process thus making statistics important the emphasis of this book is clearly on the physics of failure and developing the reliability engineering tools required for product improvements during device design and device fabrication phases NBS Special Publication **Chemical Interactions** ,2005 Catalog of National Bureau of Standards Publications, 1966-1976: Key word index United States. National Bureau of Standards. Technical Information and Publications Division, 1978 Fullerenes H. W. Kroto, D. R. M. Walton, 1993-09-09 In 1985 Buckminsterfullerene fullerene 60 C60 was discovered serendipitously during graphite laser vaporisation experiments designed to simulate the chemistry in a red giant carbon star The molecule was isolated for the first time in macroscopic amounts in 1990 a breakthrough which triggered an explosion of research into its chemical and physical properties. The fullerenes hold great potential for material science applications such as semiconductors and microscopic engineering and as new compounds for pharmaceuticals polymers and the chemical industry In October 1992 a Discussion Meeting of the Royal Society entitled A Post Buckminsterfullerene View of the Chemistry Physics and Astrophysics of Carbon organised by H W Kroto A L Makay G Turner and D R M Walton was held to celebrate this exciting advance The scientists who played key roles in the discovery and who are currently uncovering fascinating problems and the implications of this elegant molecule presented the papers published in this book

Publications of the National Bureau of Standards 1977 Catalog United States. National Bureau of Standards, 1978 Scientific and Technical Aerospace Reports, 1989 Bridge Maintenance, Safety, Management and Life Extension Airong Chen, Dan Frangopol, Xin Ruan, 2014-05-27 Bridge Maintenance Safety Management and Life Extension contains the lectures and papers presented at IABMAS 2014 the Seventh International Conference on Bridge Maintenance Safety and Management IABMAS 2014 held in Shanghai China July 7 11 2014 This set consists of a book of abstracts and a DVD containing the full papers of 396 contributions presented at IABMAS 2014 including the TY Lin Lecture nine Keynote Lectures and 386 technical papers from about 40 countries All major aspects of bridge maintenance safety management and life extension are addressed including advanced materials aging of bridges assessment and evaluation bridge codes bridge diagnostics bridge management systems damage identification design for durability deterioration modeling earthquake and accidental loadings fatigue field testing health monitoring load models life cycle assessment maintenance strategies non destructive testing prediction of future traffic demands repair and replacement residual service life safety and serviceability service life prediction and sustainable bridges among others This set provides an up to date overview of the field of bridge engineering as well as the recent significant contributions to the process of making rational decisions in bridge maintenance safety management sustainability monitoring long life strategy and resources optimization deployment for the purpose of enhancing the welfare of society It will serve as a valuable reference to all concerned with bridge structure and infrastructure systems including students researchers engineers consultants and contractors from all areas of bridge

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