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# Bloomberg Essentials Answers

**Robert N. Lussier, John R. Hendon**



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web we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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web we can express the first law in many ways one of the more useful expressions is that the change in internal energy  $\Delta E$  of a system in any process is equal to the heat  $q$  added to the system plus the work  $w$  done on the system by its surroundings  $\Delta E = q + w$

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