

Name: _____ Period: _____ Date: _____

Student Exploration: Boyle's Law and Charles' Law

Essential questions are thought-provoking questions that guide your learning.

Essential Question: How does Boyle's Law and Charles' Law work?

As you learn, think about how these laws relate to the world around you. How do they explain the behavior of gases?

Use the space below to write your answers to the essential questions as you learn. You may want to use a separate sheet of paper for your answers.

Boyle's Law

Boyle's Law states that the volume of a gas is inversely proportional to its pressure. In other words, if the pressure of a gas increases, its volume decreases, and vice versa.

1. What happens to the volume of a gas if the pressure increases?

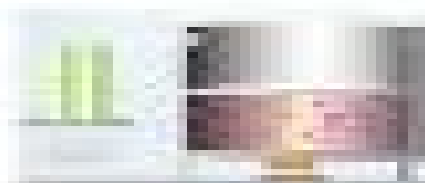
Answer: _____

2. How does the volume of a gas change if the pressure decreases?

Answer: _____

3. How does the volume of a gas change if the temperature increases?

4. How does the volume of a gas change if the temperature decreases?



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CH Cherryholmes

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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