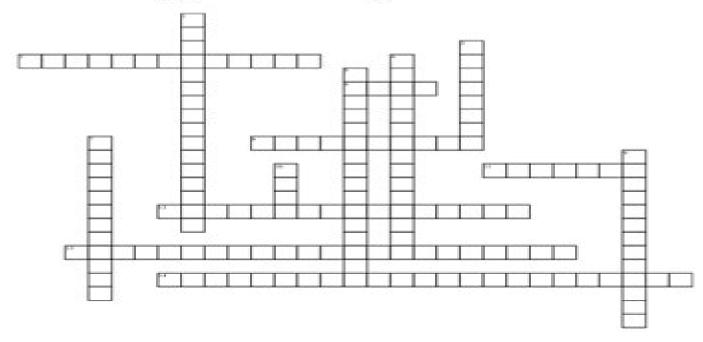
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# Bonding, Naming and Reactions



#### Across

 chemical compound composed of ions held together by electrostatic forces termed ionic bonding

a substance that donates hydrogen ions

8, a digit or symbol that appears below the line

 a logarithmic scale used to specify the acidity or basicity of an aqueous solution

 a chemical bond that involves the sharing of electron pairs between atoms  when an acid and a base react to form water and a salt and involves the combination of H+ ions and OH- ions to generate water

14. states that for any system closed to all transfers of matter and energy, the mass of the system must remain constant over time, as system's mass cannot change, so quantity cannot be added nor removed

#### Dogwo

 the symbolic representation of a chemical reaction in the form of symbols and formula

species of atom having the same number of protons in their atomic nuclei  number assigned to an element in chemical combination that represents the number of electrons lost or gained by an atom of that element in the compound

5. outer shell electron that is associated with an atom

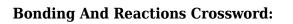
constant term related to the properties of a product

 tabular arrangement of the chemical elements, arranged by atomic number, electron configuration, and recurring chemical properties

 substances that in aqueous solution, release hydroxide ions

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**Julia McMeans** 



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