Name

Bond energy is defined as the amount of energy required to break a bond. These values are positive, indicating that bond breaking is endothermic. Bond energies are reported in kilojoules per mole (kJ/mol). The energy for breaking a hydrogen-hydrogen bond is 436 kJ/mol so when a hydrogen-hydrogen bond is formed the process releases 436 kJ/mol...

In a chemical reaction several bonds are broken and formed. For example in the reaction below a hydrogen-hydrogen bond is broken and a fluorine-fluorine bond is broken. Two hydrogen-fluorine bonds are formed. The overall energy change for this process is calculated below.

H-HBland energy: 436 kJ/mol

F-FBond Energy: 155 kJ/mol

example
$$H_2(g) + F_2(g) \rightarrow 2HF$$

$$\Delta H_m = [\text{energy used for breaking bonds}] - [\text{energy formed in making bonds}]$$

$$[436 \text{ kJ/mol} + 155 \text{ kJ/mol}] - [2(567 \text{ kJ/mol})]$$

$$= -543 \text{ kJ/mol}$$

Average Bond Energies (kJ/mol)

H-H	436 kJ/mod	C-H	413 kl/mol	C=C	614 kJ/mel
H-CI	431 kWmol	C-C	348 kJ/mol	C=C	839 kJ/mol
H-F	567 kJ/mol	C-N	293 kJ/mol	C=O	799 kJ//med
NHH	3/91 kJ/mod	C-O	3.58 kil/imoli	O=O	495 k.Wimoli
N-O	201 kJ/mol	C-F	485 kJ/mod	CHO	1072 kJ/mod
O-H	463 kJ/mol	C-CI	328 kJ/mol	CHN	6.1.5 k.Winnell
0-0	146 kJ/mol	C-S	259 kJ/mol	N-N	418 kJ//mol
F-F	155 kJ/mol	CI-CI	242 kJ/mol	No. 194	941 kJ/mol
				CHN	891 kJ/mol

Estimate the enthalpy change (ΔH_{res}) of the following reactions using the bond energies above.

1.
$$H-H+CI-CI \rightarrow H-CI+H-CI$$
4. $H-\frac{CI}{C}-CI+O \rightarrow \frac{O}{C}-CI+H-CI$
2. $H-C=C + H-F-F \rightarrow H-\frac{H}{C}-\frac{H}{C}-H$
5. $H-\frac{C}{C}-H+O=O \rightarrow O=C=O+H-O-H$
3. $H-\frac{H}{C}-O-H+H-\frac{H}{C}-O-H \rightarrow H-\frac{H}{C}-O-H+H-O-H$

Draw Lewis structures for reactants and products. Estimate the enthalpy change (ΔH_{re}) for the reactions using bond energies.

9.
$$N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$$

10.
$$H_2(g) + C_2H_4(g) \rightarrow C_2H_6(g)$$

11.
$$2C_2H_6(g) + 7O_2(g) \rightarrow 4CO_2(g) + 6H_2O(g)$$

13. HCN
$$(g)$$
 + 2H₂ (g) \rightarrow CH₃NH₂ (g)

Bond Energy Extension Questions

Eileen Ramsden

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understanding presupposes that there exists a body of theory capable of explaining situations not easily accessible to experimental methods and of relating experimental findings among themselves and with theoretical constructs For these reasons theoretical developments in surface physics and surface chemistry of transition metal compounds have been of paramount importance in promoting progress in catalysis electronic devices corrosion etc Although a great variety of spectroscopic methods for analyzing solids and surfaces at molecular scale have been introduced in recent years nevertheless many questions about the adsorption sites and intermediates the effect of promoters the poisoning of active sites the nature of segregation of impurities the process of surface reconstruction the mechanisms of reactions etc have remained unanswered simply because of the great complexity of surface phenomena It is in this sense that quantum mechanical method combined with experimental data may shed some light on the microscopic properties of new surface Adhesive Bonding L.H. Lee, 2013-06-29 For several years I have been responsible for organizing and teaching in the fall a short course on Fundamentals of Adhesion Theory Practice and Applications at the State University of New York at New Paltz Every spring I would try to assemble the most pertinent subjects and line up several capable lecturers for the course However there has always been one thing missing an authoritative book that covers most aspects of adhesion and adhesive bonding Such a book would be used by the participants as a main reference throughout the course and kept as a sourcebook after the course had been completed On the other hand this book could not be one of those All you want to know about volumes simply because adhesion is an interdisciplinary and ever growing field For the same reason it would be very difficult for a single individual especially me to undertake the task of writing such a book Thus I relied on the principle that one leaves the truly monumental jobs to experts and I finally succeeded in asking several leading scientists in the field of adhesion to write separate chapters for this collection Some chapters emphasize theoretical concepts and others experimental techniques In the humble beginning we planned to include only twelve chapters However we soon realized that such a plan would leave too much ground uncovered and we resolved to increase the coverage After the book had evolved into thirty chapters we started to feel that perhaps our mission had been accomplished Atomic Industry Reporter ,1959

High Entropy Materials Krishanu Biswas, Nilesh Prakash Gurao, Tanmoy Maiti, Rajiv S. Mishra, 2022-11-22 This book provides a detailed overview of high entropy materials and alloys discussing their structure the processing of bulk and nanostructured alloys as well as their mechanical and functional properties and applications. It covers the exponential growth in research which has occurred over the last decade discussing novel processing techniques estimation of mechanical functional and physical properties and utility of these novel materials for various applications. Given the expanding scope of HEAs in ceramics polymers thin films and coating this book will be of interest to material scientists and engineers alike

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